

Weavers Academy
SCIENCE
KS5 Curriculum Maps



BIOLOGY



Year 12: Curriculum Map

Term	Content Covered
Autumn 1	Foundations in Biology <ul style="list-style-type: none">• Cell structure• Cell cycle and division
Autumn 2	Foundations in Biology <ul style="list-style-type: none">• Biological molecules• Nucleic acids• Enzymes• Biological membranes
Spring 1	Exchange and Transport <ul style="list-style-type: none">• Exchange surfaces
Spring 2	Exchange and Transport <ul style="list-style-type: none">• Transport in animals• Transport in plants
Summer 1	Biodiversity, evolution and disease <ul style="list-style-type: none">• Communicable diseases• Biodiversity
Summer 2	Biodiversity, evolution and disease <ul style="list-style-type: none">• Classification

BIOLOGY



Year 13: Curriculum Map

Term	Content Covered
Autumn 1	<p>Communication, Homeostasis, and Energy Communication and homeostasis</p> <ul style="list-style-type: none">• The need for communication• Homeostasis• Temperature regulation in endo and ectotherms <p>Excretion as an example of homeostatic control</p> <ul style="list-style-type: none">• Excretion• Liver structure and function• Kidney structure and function• Osmoregulation• Kidney failure <p>Neuronal control</p> <ul style="list-style-type: none">• Roles of sensory receptors• Structure and function of neurones• Nerve impulses• Synapses <p>Hormonal control</p> <ul style="list-style-type: none">• Endocrine communication• Adrenal glands• The pancreas and insulin• Regulation of blood sugar levels• Diabetes

Autumn 2	<p>Plant and Animal responses</p> <ul style="list-style-type: none"> • Plant responses to the environment • Plant growth • Tropisms • Commercial use of plant hormones • Mammalian nervous system • The brain • Reflex actions • Coordinating response • Controlling the heart • Muscles and muscular contraction. <p>Photosynthesis</p> <ul style="list-style-type: none"> • Relationship between photosynthesis and respiration • Chloroplasts and photosynthetic pigments • Light independent stage • Light dependent stage • Factors affecting photosynthesis <p>Respiration</p> <ul style="list-style-type: none"> • The need for cellular respiration • Glycolysis • Structure of mitochondria • Link reaction and Krebs's cycle • Oxidative phosphorylation and chemiosmotic theory • Anaerobic respiration in eukaryotes • Factors affecting the rate of respiration <p>Respiratory substrates</p>
Spring 1	<p>Genetics and Ecosystems Cellular control</p> <ul style="list-style-type: none"> • Gene mutations and regulation of gene expression • Genetic control of body plan

	<p>Patterns of inheritance</p> <ul style="list-style-type: none"> • Genetic variation • Mono and Dihybrid inheritance • Multiple alleles and sex linkage • Codominance • Autosomal linkage • Epistasis • Chi-squared • Continuous and discontinuous data • Factors effecting evolution • Hardy-Weinberg • Isolation mechanisms • Artificial selection <p>Manipulating Genomes</p> <ul style="list-style-type: none"> • DNA sequencing • Applications of gene sequencing • DNA profiling • The PCR • Electrophoresis • Genetic engineering • Issues relating to genetic manipulation • Gene therapy <p>Cloning and Biotechnology</p> <ul style="list-style-type: none"> • Natural clones • Clones in plants • Artificial clones in animals • Biotech and food • Microorganism cultures and Population growth • Immobilised enzymes
Spring 2	<p>Ecosystems</p> <ul style="list-style-type: none"> • Ecosystems • Transfer and manipulation of biomass • Succession

	Populations and sustainability <ul style="list-style-type: none"> • What determines population size • Interactions between populations • Conservation and preservation • Sustainable management • Balancing the conflict between conservation and human needs • Controlling the effects of human activities
Summer 1	Revision

CHEMISTRY



Year 12: Curriculum Map

Term	Content Covered
Autumn 1	Atomic Structure, Atoms and Reactions <ul style="list-style-type: none">• The development of the atomic model and the associated chronology• Atomic structure and isotopes• The importance of quantification Chemistry in the real world• Composition of balanced symbol equations (including ionic equations)• Mr, Mass, Mole calculations to include gas volumes and concentrations• % yield and atom economy calculations• Titration calculations• Acids, bases, and neutralisation• Oxidation numbers and Redox reactions
Autumn 2	Electrons, Bonding and Structure <ul style="list-style-type: none">• Energy levels, shells, sub-shells, atomic orbitals, electron configuration• Ionic Bonding• Covalent Bonding• Composition of Dot and Cross Diagrams• Metallic Bonding• Electronegativity and bond polarity• Intermolecular forces• Shapes of Molecules
Spring 1	The Periodic Table <ul style="list-style-type: none">• The structure of the Periodic Table and the 'story' of how it evolved• Periodic trend in electron configuration and ionisation energy• Periodic trend in structure and melting point and boiling points

	<ul style="list-style-type: none"> Identifying the bonding and structure in given compounds by applying rules associated with the types of atoms present Redox reactions and reactivity of Group 2 metals Reactions of Group 2 compounds Characteristic physical properties of the Halogens Characteristic reactions of halide ions Tests for ions
Spring 2	Energy <ul style="list-style-type: none"> Construction of Energy profile diagrams Enthalpy changes: ΔH of reaction, formation, combustion and neutralisation Measurement of temperature Determination of Enthalpy changes of neutralisation and combustion Bond enthalpies and associated calculations Hess' law and enthalpy cycles Collision Theory Catalysis The Boltzmann Distribution Dynamic equilibrium and le Chatelier's principle The equilibrium constant, K_c
Summer 1	Basic Organic Concepts and Hydrocarbons <ul style="list-style-type: none"> Naming and representing the formulae of organic compounds Functional groups Application of IUPAC rules in developing a systematic framework for chemical nomenclature Use of the general formula of a homologous series to predict the formula of any member of the series Use of the 'curly arrow' model to demonstrate electron flow in organic reactions Use of Cahn–Ingold–Prelog (CIP) priority rules to identify the E and Z stereoisomers Use of Markownikof's rule to predict formation of a major organic product in addition reactions of H–X to unsymmetrical alkenes Isomerism Properties of alkanes Reactions of alkanes Properties of alkenes Stereoisomerism in alkenes

	<ul style="list-style-type: none"> • Addition reactions of alkenes • Polymers from alkenes • Waste polymers and alternatives
Summer 2	Alcohols, Haloalkanes and Analysis <ul style="list-style-type: none"> • Properties of alcohols • Reactions of alcohols • Substitution reactions of haloalkanes • Free radical equations • Environmental concerns from use of organo-halogen compounds • Synthetic routes • Infrared spectroscopy • Mass spectrometry • Combined analytical techniques

CHEMISTRY



Year 13: Curriculum Map

Term	Content Covered
Autumn 1	Rates, Equilibrium and pH <ul style="list-style-type: none">• Orders, rate equations and rate constants• Rate graphs and orders• Rate-determining step• Effect of temperature on rate constants• Equilibrium• Brønsted–Lowry acids and bases• pH and $[H^+(aq)]$• Buffers: action, uses and calculations• Neutralisation
Autumn 2	Energy <ul style="list-style-type: none">• Lattice enthalpy• Born–Haber and related enthalpy cycles• Entropy• Free energy• Redox• Electrode potentials and Electrochemical Cells• Storage and fuel cells Transition Metals <ul style="list-style-type: none">• Transition metal properties• Ligands and complex ions• Ligand substitution• Precipitation reactions• Redox reactions

Spring 1	Organic Chemistry <ul style="list-style-type: none"> • Benzene and aromatic compounds • Electrophilic substitution • Phenols • Reactions of carbonyl compounds • Characteristic tests for carbonyl compounds • Properties of carboxylic acids • Esters • Acyl chlorides • Basicity and preparation of amines • Reactions of amino acids • Amides • Chirality • Condensation polymers • Extending carbon chain length • Stereoisomerism • Synthetic routes
Spring 2	Analysis <ul style="list-style-type: none"> • Types of chromatography • Tests for organic functional groups • Tests for organic functional groups • GC and TLC • (FT-)IR • Proton and Carbon-13 NMR Spectroscopy • Combined techniques: deduction of the structures of organic compounds from different analytical data
Summer 1	Revision

PHYSICS



Year 12: Curriculum Map

Term	Content Covered
Autumn 1	<p>Motion</p> <ul style="list-style-type: none">• Kinematics• graphs of motion• SUVAT• projectile motion• stopping distances <p>Charge and current</p> <ul style="list-style-type: none">• Circuit components• current and charge• electron drift• velocity <p>Energy and power</p> <ul style="list-style-type: none">• PD and emf• Ohm's law• Resistivity and temperature• Power <p>Cost of electricity</p>
Autumn 2	<p>Forces in action</p> <ul style="list-style-type: none">• Newtons• Dynamics• Equilibrium• Turning forces• Centre of mass• Density

	<ul style="list-style-type: none"> • Pressure <p>Electric circuits</p> <ul style="list-style-type: none"> • Kirchoff's laws • Series and parallel circuits • Potential dividers • Internal resistance • Circuit analysis
Spring 1	<p>Work, energy power</p> <ul style="list-style-type: none"> • Work and the joule • Conservation of energy • Potential and kinetic energy • Power and the Watt • Efficiency <p>Materials</p> <ul style="list-style-type: none"> • Deformation of materials • The Young Modulus • Categorisation of materials <p>Newtons laws</p> <ul style="list-style-type: none"> • Newton's 1st, 2nd and 3rd law • Momentum • Force & Impulse • Elastic and inelastic collisions <p>Wave motion</p> <ul style="list-style-type: none"> • Wave terminology • Wave speed and wave equation • Properties of waves <p>Wave superposition</p> <ul style="list-style-type: none"> • Interference • Young double-slit experiment • Diffraction gratings

Spring 2	<p>Electromagnetic waves</p> <ul style="list-style-type: none"> • EM waves • Polarisation • Refraction • TIR <p>Stationary waves</p> <ul style="list-style-type: none"> • Stationary waves • Experiments with SW <p>Stationary longitudinal waves</p>
Summer 1	<p>Quantum physics</p> <ul style="list-style-type: none"> • The photon • The electronvolt • The photoelectric effect • Wave-particle duality
Summer 2	<p>Nuclear and particle physics</p> <ul style="list-style-type: none"> • Nuclear atom • strong nuclear force • nuclear density <p>Astrophysics</p> <ul style="list-style-type: none"> • Structure of the universe • Star life cycle • EM radiation from stars • Wien's and Stefan's law, Astronomical distances • Doppler effect • CMBR • Expansion of the Universe • Dark matter and dark energy

PHYSICS



Year 13: Curriculum Map

Term	Content Covered
Autumn 1	<p>Thermal physics</p> <ul style="list-style-type: none">• Temperature• Solids, liquids and gases• Internal energy• Brownian motion• Specific heat capacity• Specific latent heat• Kinetic theory• Investigating gases• Ideal gas equation• The Boltzmann constant <p>Circular motion</p> <ul style="list-style-type: none">• Kinematics• Centripetal force <p>Oscillations</p> <ul style="list-style-type: none">• SHM• Equations of SHM• Graphical analysis of SHM• Simple harmonic oscillator• Damping
Autumn 2	<p>Gravitational fields</p> <ul style="list-style-type: none">• Newton's law of gravitation• Motion of planets and satellites• Gravitational potential and GPE

	<p>Capacitors</p> <ul style="list-style-type: none"> • Capacitors in series and parallel circuits • Energy, Charging and discharging <p>Electric fields</p> <ul style="list-style-type: none"> • Coulomb's law • Uniform electric fields • Electric potential and EPE <p>Electromagnetism</p> <ul style="list-style-type: none"> • Magnetic fields • Magnetic flux density • Forces on current-carrying wires • Motion of a charged particle in E and B fields • Electromagnetic induction • Faraday's and Lenz's laws • AC generators • Transformers
Spring 1	<p>Nuclear and particle physics</p> <ul style="list-style-type: none"> • Fundamental particles • Radioactivity • Radioactive decay • Half-life • Radioactive dating • Mass-energy conservation • Nuclear fission • Nuclear fusion <p>Medical imaging</p> <ul style="list-style-type: none"> • X-rays • Attenuation of X-rays • CAT scans • Gamma camera • PET scanning • Ultrasound • Acoustic impedance

	<ul style="list-style-type: none">• Doppler effect
Spring 2	Revision
Summer 1	Revision

MEDICAL SCIENCES



Year 12: Curriculum Map

Term	Content Covered
Autumn 1	<p>Classes of biological molecules</p> <ul style="list-style-type: none">• Carbohydrates• Lipids• Proteins and enzymes• Mechanisms of action• Nucleotides• Nucleic acids <p>Factors affecting enzyme reactions</p> <p>Lifestyle and Health</p>
Autumn 2	<p>Human cellular structure</p> <ul style="list-style-type: none">• Plasma membrane• Nucleus• Nucleolus• Endoplasmic reticulum• Golgi apparatus• Mitochondria• Nuclear envelope <p>Movement into and out of cells</p> <ul style="list-style-type: none">• Simple diffusion• Osmosis• Facilitated diffusion• Active transport• Endo/exocytosis

	DNA mechanisms <ul style="list-style-type: none"> • Semi conservative replication • Transcription • Translation
Spring 1	Human Physiological Systems <ul style="list-style-type: none"> • Describe structure of human physiological systems • Explain function of human physiological systems • Explain principles of physiological measurement tests • Describe limitations of physiological measurement testing
Spring 2	Infection and Response <ul style="list-style-type: none"> • Explain how pathogens can affect body systems • Explain how non-infectious diseases affect body systems
Summer 1	Revision Unit 1
Summer 2	Medicines and Treatment Options <ul style="list-style-type: none"> • Explain factors to be considering when prescribing medicines • Strategies to improve adherence of patients taking prescriptions • Compare options for administering medicines • Explain the molecular basis of the action of medicines • Explain how medicines affect body systems • Explain how medicines affect causative agents of infectious disease • Explain why medicines may lose their effectiveness • Compare the effects of the interaction of medicines • Explain how factors affect the distribution of medicines in the body • Explain how adverse reactions to medicines can occur • Explain fate of medicines in the body • Describe what is meant by the term cancer • Explain the genetic basis of cancer • Describe possible treatment options for cancer • Assess the potential impact of new treatments for cancer

MEDICAL SCIENCES



Year 13: Curriculum Map

Term	Content Covered
Autumn 1	Clinical Laboratory Techniques and Assessment of Biological samples using clinical tests <ul style="list-style-type: none">• Explain factors that affect clinical test results• Plan tests• Record results from tests• Be able to process data from clinical tests• Use graphs to process data• Use numerical methods to process data• Interpret data from clinical tests• Communicate information to an audience
Autumn 2	Clinical Laboratory Techniques and Assessment of Biological samples using clinical tests <ul style="list-style-type: none">• Explain factors that affect clinical test results• Plan tests• Record results from tests• Be able to process data from clinical tests• Use graphs to process data• Use numerical methods to process data• Interpret data from clinical tests• Communicate information to an audience
Spring 1	Physiological Measurements, Diagnosis, Treatment and Medical Research <ul style="list-style-type: none">• Understand physiological information presented within case studies• Understand how physiological measurement techniques can be used to support diagnosis and treatment• Understand how medical research can help support diagnosis and treatment

Spring 2	Physiological Measurements, Diagnosis, Treatment and Medical Research <ul style="list-style-type: none"> • Understand ways in which medical treatments can be used to treat diseases and disorders • Understand ways in which clinical measurement techniques can be used to support diagnosis and treatment
Summer 1	Revision for assessment